

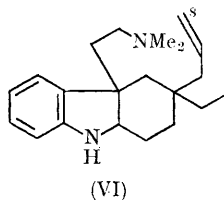
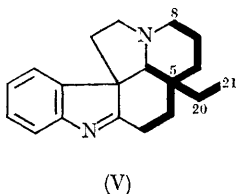
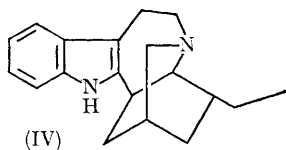
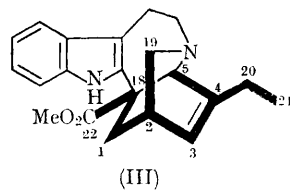
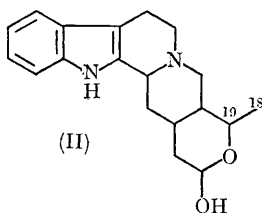
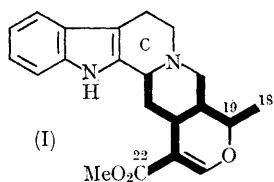
Biosynthesis of the Indole Alkaloids

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Our previous researches^{1,2} eliminated two of the four biogenetic schemes³ which had been proposed to account for the origin of the C₉₋₁₀ unit of the indole and Ipecacuanha alkaloids [thickened bonds in (I), (III), and (V)]. A third scheme, based upon a one-carbon unit, acetic and malonic acids, had received experimental support⁴ but this too has been firmly excluded.^{1,2,5-7} Importantly, the proof that a one-carbon unit is *not* involved in the biosynthesis^{1,2,5,6} directed attention to the fourth theory⁸ which suggested a relationship of the C₉₋₁₀ unit to the cyclopentanoid monoterpene skeleton (VIII). Pointers in favour of this idea came from the incorporation of labelled mevalonic acid into various indole^{9,5,2} and Ipecacuanha² alkaloids. We now report degradative results¹⁰ which show the mevalonoid origin of representative

(I) to yield (II) and from dihydrocatharanthine to give epi-ibogamine¹² (IV) showed by the changes in molar activities that C-22 of (I) and (III) carry, respectively, 24% and 23% of the original total activity. These results establish the specific incorporation of mevalonic acid into the alkaloids (I) and (III) in a way consistent with the annexed scheme.⁹ In this, C-2 and C-6 of one mevalonate unit have been rendered equivalent as observed for plumeride¹³ (see also below). Kuhn-Roth oxidation of (III) gave inactive propionic acid and dehydrogenation of (III) over palladium¹² yielded 3-ethylpyridine (48% of total activity). A partial assignment of the radioactivity in (III) can thus be made: C-22, 23%; C-1 + C-18 by difference, 29%; and 48% located at some point(s) in C-2,-3,-5 and -19. These results are entirely consistent with the



examples from the three major classes of indole alkaloids and give insight into the mode of linkage of the two C₅ units.

Sodium (\pm)-[2-¹⁴C]mevalonate (VII) was incorporated by *Vinca rosea* plants to yield radioactive vindoline (0.05% incorporation), ajmalicine (I; 0.003% incorporation), catharanthine (III; 0.04% incorporation) and serpentine (I; ring-c aromatised; 0.02% incorporation). The *O*-methyl groups of (I) and (III) were shown to carry no significant radioactivity by Zeisel determination. Both alkaloids were further proved to be radiochemically pure by dilution and by conversion into a suitable derivative of unchanged molar activity. Elimination of the methoxycarbonyl group¹¹ from

scheme shown (requires C-22, 25%; C-1, 25%; C-19, 50%).

Rhazia stricta plants similarly incorporated sodium (\pm)-[2-¹⁴C]mevalonate to afford radioactive 1,2-dehydroaspidospermidine¹⁴ (V, 0.15% incorporation) which was reduced to aspido-spermidine.¹⁴ Kuhn-Roth oxidation established that C-5, -20, and -21 carried no radioactivity. Further degradation by Emde and Hofmann steps¹⁵ to the base (VI) allowed the isolation of C-8 as formaldehyde (65 \pm 6% of total). Since the starred carbon in (XI) is lost during the formation of (V), the expected value for C-8 (of V) is 67% of the total activity in the complete absence of scatter. The same alkaloid isolated from plants

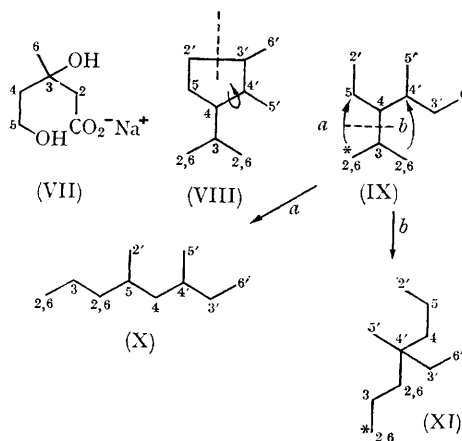
fed with sodium (\pm)-[3- 14 C]mevalonate (0.15% incorporation) was similarly reduced and the product oxidised (Kuhn-Roth) to acetic acid (47% of total activity) and propionic acid (47% of total). Schmidt degradation of the acetic acid gave inactive methylamine. It follows that C-20 of (V) carries 47% of the original activity whilst C-5 and -21 are essentially radio-inactive.

Independent studies by Arigoni and Scott and their respective co-workers have been in progress concurrently. They showed that vindoline^{16,17} and reserpine¹⁷ derived from [2- 14 C]mevalonic acid carry, respectively, 22% and 26% of the total activity at the positions corresponding to the starred carbon atom in (IX) and (XI). There is thus exact agreement between their findings and the corresponding portion of the work outlined above.

All our results are in keeping with the illustrated scheme and they eliminate the recent alternative proposal¹⁶ which requires labelling patterns for (III) and (V) differing from those found.

Experiments with 4-, 5-, and 6-labelled mevalonic

acid are in progress to gain further knowledge of the biosynthesis.



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¹ A. R. Battersby, R. Binks, W. Lawrie, G. V. Parry, and B. R. Webster, *Proc. Chem. Soc.*, 1963, 369.

² A. R. Battersby, R. Binks, W. Lawrie, G. V. Parry, and B. R. Webster, *J. Chem. Soc.*, 1965, 7459.

³ Reviewed by A. R. Battersby, "Biogenesi delle Sostanze Naturali", Accademia Nazionale dei Lincei, Roma, 1964, p. 47.

⁴ P. N. Edwards and E. Leete, *Chem. and Ind.*, 1961, 1666; E. Leete, S. Ghosal, and P. N. Edwards, *J. Amer. Chem. Soc.*, 1962, **84**, 1068; E. Leete and S. Ghosal, *Tetrahedron Letters*, 1962, 1179.

⁵ H. Goeggel and D. Arigoni, *Experientia*, 1965, **21**, 369.

⁶ D. H. R. Barton, G. W. Kirby, R. H. Prager, and E. M. Wilson, *J. Chem. Soc.*, 1965, 3990.

⁷ E. Leete, A. Ahmad, and I. Kompis, *J. Amer. Chem. Soc.*, 1965, **87**, 4168.

⁸ R. Thomas, *Tetrahedron Letters*, 1961, 544; E. Wenkert, *J. Amer. Chem. Soc.*, 1962, **84**, 98.

⁹ T. Money, I. G. Wright, F. McCapra, and A. I. Scott, *Proc. Nat. Acad. Sci., U.S.A.*, 1965, **53**, 901.

¹⁰ Reported in part at the Chemical Society Symposium, "Organic Approaches to Biosynthesis" Imperial College, London, 21st October, 1965.

¹¹ E. Wenkert and N. V. Bringi, *J. Amer. Chem. Soc.*, 1959, **81**, 1474.

¹² N. Neuss and M. Gorman, *Tetrahedron Letters*, 1961, 206.

¹³ D. A. Yeowell and H. Schmid, *Experientia*, 1964, **20**, 250.

¹⁴ G. F. Smith and M. A. Wahid, *J. Chem. Soc.*, 1963, 4002.

¹⁵ Cf. H. Conroy, P. R. Brook, M. K. Rout, and N. Silverman, *J. Amer. Chem. Soc.*, 1958, **80**, 5178; H. Conroy, P. R. Brook, and Y. Amiel, *Tetrahedron Letters*, 1959, No. 11, 4.

¹⁶ F. McCapra, T. Money, A. I. Scott, and I. G. Wright, *Chem. Comm.*, 1965, 537.

¹⁷ H. Goeggel and D. Arigoni, *Chem. Comm.*, 1965, 538.